## EFFECTS OF ELECTRODEPOSITED LITHIUM PHOSPHATE ON THE ELECTROCHEMICAL PROPERTIES OF SELF ORGANIZED TITANIA NANOTUBES

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Solid electrolyte interfaces are crucial for the development of lithium-ion batteries. An alternative approach is to coat the electrodes with a convenient ionic conductor that warrants lithium ion diffusion from the electrolyte. Thin film electrodes prepared directly onto current collectors are particularly suitable to this approach. The compound Li<sub>3</sub>PO<sub>4</sub> is known to be a potential candidate for thin film lithium electrolyte due to its light weight, wide potential range of stability vs. lithium and also because of its thermal and mechanical stability.<sup>1</sup>

The aim of this work is to prepare nanoarchitectured electrodes containing a thin layer of Li<sub>3</sub>PO<sub>4</sub>. The covering effect of Li<sub>3</sub>PO<sub>4</sub> can be observed for 1 min of electrodeposition at 37.5 mA cm<sup>-2</sup> as the nanotubes are filled (Fig. 1). Firstly, self organized TiO<sub>2</sub> nanotubes are used to improve their electronic and ionic conductivity. The annealing condition allowed preparing  $\beta$ -Li<sub>3</sub>PO<sub>4</sub> (R.T. – 300 °C) and  $\gamma$ -Li<sub>3</sub>PO<sub>4</sub> ( $\ge$  500 °C).<sup>2</sup> The discharge and charge plateau of Li<sub>3</sub>PO<sub>4</sub>-free ntTiO<sub>2</sub> samples were at approximately 1.75 and 1.9 V, respectively. For  $\beta$ -Li<sub>3</sub>PO<sub>4</sub>/ntTiO<sub>2</sub> and  $\gamma$ -Li<sub>3</sub>PO<sub>4</sub>/ntTiO<sub>2</sub> samples deposited 1 min at 3.75 mA cm<sup>-2</sup> a discharge/charge plateaus are observed at 1.78 and 1.86 V, which is ~0.1 V lower in energy as compared with uncoated samples. The best capacity value of 500  $\mu$ A h cm<sup>-2</sup> was achieved for TiO<sub>2</sub>/ $\gamma$ -Li<sub>3</sub>PO<sub>4</sub>.

The optimum parameters for electrodeposition to achieve  $\gamma$ -Li<sub>3</sub>PO<sub>4</sub> coating on the entire ntTiO<sub>2</sub> array are 1 min and 3.75  $\mu$ A cm<sup>-2</sup>. The performance of the ntTiO<sub>2</sub>/ $\gamma$ -Li<sub>3</sub>PO<sub>4</sub> (LiPF<sub>6</sub> in EC:DEC)/ LiFePO<sub>4</sub> rocking-chair microbattery delivered a maximum capacity of 110 mA h g<sup>-1</sup> at 5C rate when imposing a cathode-limited active mass ratio to the electrodes.<sup>2</sup> The obtained low-voltage microbattery (<2.0 V) could fit to applications demanding small-scale electrical power.

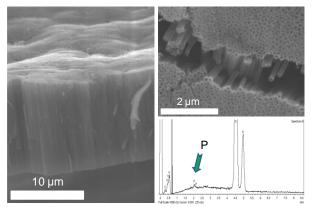


Figure 1. SEM image of nt-TiO<sub>2</sub> with electrodeposited Li<sub>3</sub>PO<sub>4</sub>.

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<sup>&</sup>lt;sup>1</sup> Bates, J. B.; Dudney, N. J.; Gruzalski, G. R.; Zuhr, R. A.; Choudhury, A.; Luck, D. F.; Robertson, J. D. *J. Power Sources* **1993**, *43*, 103.

<sup>&</sup>lt;sup>2</sup> López, M.C.; Ortiz, G.F.; González, J.R.; Alcántara, R.; Tirado, J.L. ACS Appl. Mater. Interfaces **2014**, *6*, 5669.