




Review

# Nitrogen Fertilization. A Review of the Risks Associated with the Inefficiency of Its Use and Policy Responses

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**Abstract:** Nitrogen (N) is a key input to food production. Nearly half of N fertilizer input is not used by crops and is lost into the environment via emission of gases or by polluting water bodies. It is essential to achieve production levels, which enable global food security, without compromising environmental security. The N pollution level expected by 2050 is projected to be 150% higher than in 2010, with the agricultural sector accounting for 60% of this increase. In this paper, we review the status of the pollution from N fertilizers worldwide and make recommendations to address the situation. The analysis reviews the relationship between N fertilizer use, N use efficiency, non-point pollution, the role of farmer management practices, and policy approaches to address diffuse pollution caused by N fertilization. Several studies show a lack of information as one of the main hurdles to achieve changes in habits. The objective of this study is to highlight the gravity of the current global non-point pollution as well as the need for a communication effort to make farmers aware of the relationship between their activity and N pollution and, therefore, the importance of their fertilizer management practices.

**Keywords:** review; nitrogen pollution; nitrogen-use efficiency; eutrophication; lixiviation; volatilization



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## 1. Introduction

Nitrogen (N), carbon, hydrogen, oxygen, and phosphorus are the elements on which life is based [1]. Among them, N is the fundamental element in the formation of amino acids, which, in turn, are the basic components of proteins. This element makes up 78% of the air in the atmosphere where it is practically inert and largely unreactive, existing mostly in molecular form (N<sub>2</sub>). Nitrogen gas in the atmosphere is unsuitable for uptake by living matter. Human protein intake depends on the capacity of plants to take N into the food chain, and it requires certain transformations to be used by plants and later by animals for this to happen. Robertson and Groffman [2] defined the N cycle and its different forms. Atmospheric dinitrogen gas is turned into ammonia (NH<sub>3</sub>) and fixed in the soil through the fixation process.

The N fixation relies on a series of microbial processes that transforms N<sub>2</sub> into organic forms which then remain in the soil [3,4]. Biological fixation depends on a small range in water stress and aeration levels of the environment [5], so under natural conditions, the amounts of N fixed in soil is small [6,7]. Through the process of mineralization and nitrification, heterotrophic microorganisms transform this organic N from the soil into inorganic structures (NH<sub>4</sub><sup>+</sup> and NO<sub>3</sub><sup>-</sup>) that can readily be assimilated by plants [8]. These ions are the reactive forms of N available for biosynthesis, thus initiating the process of amino acid and protein formation. Due to the low rates of natural N cycle fixation, mineralization, and nitrification, N is the most limiting element in terrestrial ecosystems of all the soil nutrients needed for plant growth [9]. The emergence of synthetic N fertilizers

from 1913 onwards triggered the development of the Green Revolution which, in turn, helped exponentially increase the world's population over the past century [10].

Nitrogen fertilizers used in agriculture increased from 10 Tg N ha<sup>-1</sup> (1961) to 77 Tg N ha<sup>-1</sup> (2016) [11,12]. Nearly half of the N fertilizer supplied is not used by crops and is lost to the ecosystem through volatilization, run-off, or leaching [13]. These losses lead to environmental problems, such as the release of greenhouse gases, pollution of water bodies, soil acidification, or biodiversity reduction. The atmospheric level of N pollution is expected by, 2050, to be in the range 102–156% higher than in 2010 with the agricultural sector accounting for 60% of this increase [12,14].

The concept of N use efficiency (NUE), as the biological capacity of the plant to assimilate N, has been the subject of numerous previous reviews [15–19]. In this paper, we focus on reviewing the NUE as an anthropomorphic activity, and the negative consequences of N over-application in agriculture. The aim of this is to enhance the dissemination of good agricultural management practices.

This paper presents a systematic review of knowledge related to diffuse N pollution, best management practices, and improved farmer behavior. Current policies for managing diffuse pollution were initiated in the 1980s. We started our search in 1980, looking at the available academic and grey literature researching the factors which affect farmers management practices in relation to their capacity to improve diffuse pollution caused by their agricultural activity.

## 2. Impact of Excess N in Agriculture

Farmers in the developing world tend to over-fertilize [20,21] because of the uncertain of variables such as weather and soil conditions. Management practices, based on the “fertilize for good years” rule, are consistent with maximizing expected returns under yield uncertainty, as the cost of over-application is less than the opportunity cost of under-application [22]. The response of crops to fertilizer addition has often been modeled as a linear and plateau response (LRP) function that increases linearly until maximum yield is achieved, where the increase stops. This is assumed to be the standard agronomic model for response to water [23] and fertilizer, especially for N, where Liebig's law of minimum performs very well [24,25]. Due to the low NUE of the crops and the extreme mobility of the reactive forms of N in either the gas or the soluble phase, this excess N applied has a high risk to be lost to the environment [2,26–31]. The main characteristic of pollution of agricultural origin is its diffuse nature since its source cannot be easily traced [32]. The high volatility of ammonia and N oxides allows them to be transported through the atmosphere, and the high solubility of nitrate means that it is easily leached from the soil profile into aquatic ecosystems [13].

### 2.1. Atmosphere Pollution

Air pollution has direct consequences on human health. In 2017 five million people died from diseases related to air pollution, and the USA and China are among the 10 countries with the highest mortality rate [33]. Agricultural activities around the world are responsible for almost 15 percent of annual greenhouse gas emissions [34,35] and for 60–80% of global N<sub>2</sub>O emissions from human activity [36,37]. The increasing use of N fertilizers in food production worldwide has resulted in the raising of N<sub>2</sub>O concentrations in the atmosphere, where it remains as a greenhouse gas 298 times more potent than carbon dioxide [38]. Current N<sub>2</sub>O concentrations are starting to exceed the levels projected in most emission scenarios by the International Panel on Climate Change (IPCC) [39,40] and are not compatible with the climate objectives of the Paris Agreement. Tian et al. [41] have conducted the first global assessment of nitrous oxide as one of the main greenhouse gases responsible for global warming, by comprehensively quantifying its global sources and sinks. According to their study, nitrous oxide is the third most important greenhouse gas after carbon dioxide and methane. Agricultural activity is the main cause of N<sub>2</sub>O pollution as it accounts for 70% of N<sub>2</sub>O emissions in the period between 2007 and 2016. Numerous stud-

ies establish linear correlations between N fertilizer use in agriculture (including organic amendments) and N<sub>2</sub>O emissions worldwide [42–44]. However, several recent studies in corn cropping systems have reported an exponential relationship between them [45,46]. The current global mean value of N<sub>2</sub>O, emitted by N fertilizers, is close to 0.9% of the N applied in cropland according to IPCC estimates [47]. This means that every 100 kg of N fertilizer used in agriculture emits 1.0 kg of N in N<sub>2</sub>O form into the atmosphere [45]. In 2018, N<sub>2</sub>O emissions from agriculture to the atmosphere were 7.718 Gg [12].

## 2.2. Aquatic Ecosystems Pollution

Nitrate concentration is a critical indicator for measuring the level of pollution in aquatic ecosystems, as it is the most stable form of N, and its high presence disrupts the vital balance of aquatic ecosystems [48,49]. The analysis of water pollution received an important boost from the studies by Kohl et al. [50], which in 1973 used N isotopes ( $\delta^{15}\text{N}$ ) to analyze the influence of chemical fertilizers on the pollution of water bodies. Because of its effectiveness, this technique has been widely used to detect sources of nitrates in this environment [51,52].

Agriculture has been recognized as a major source of nitrate leaching pollution [53–56]. Nitrogen leaching is not only related to the amount of fertilizer applied, but also to other variables such as the timing and place of fertilizer applied in relation to the crop's growing cycle, irrigation regimes, type of fertilizer, agronomic practices, crop rotation, soil characteristics, type of soil cover, climate, etc. [57–63]. A sustainable population growth is not compatible with the degradation of groundwater reservoirs. Groundwater is increasingly being used as a source to meet growing water needs. Currently, groundwater accounts for 43% of irrigation water and more than 50% of domestic consumption worldwide [64]. Therefore, contamination of this resource seriously threatens the health of people who depend on its supply. Numerous studies show that continuous consumption of water with nitrate levels above 50 mg/L increases the likelihood of developing certain types of cancer [65–67].

Freshwater eutrophication can result from discharge of the nutrient-overloaded aquifer to a surface body. Eutrophication occurs when high concentrations of dissolved nutrients (especially N and phosphorus) meet stagnant aquatic ecosystems with restricted water input. Most eutrophication processes are caused by anthropogenic activities (agricultural fertilization, urban effluents, timber industry, recreational use of water resources, etc.). Currently, the main non-point source of high nutrient concentrations in freshwater is agricultural fertilization [68,69]. Grizzetti et al. [70] estimate the source of nitrogen at water bodies at European scale, finding a high variability, ranging from 90% to 10%, as a function of the location and the level of urban wastewater treatment. The excess of inorganic compounds alters the ecosystem balance, reducing its biodiversity and giving to the cyanobacteria an opportunity to grow in a disorderly manner and end up collapsing the habitat [71,72]. The first studies of the toxins produced by cyanobacterial populations in contaminated aquatic environments began in the 1990s [73,74]. Cyanobacteria settle in the surface area of the water body, where the incidence of solar radiation is greatest, turning the water green and preventing light from reaching deeper areas [75]. Vegetation and many floating algae die when they fall below the new threshold of photosynthesis, due to the decrease in nutrients, where they are decomposed by bacteria that, in their activity, consume oxygen and produce toxins. Native aquatic species disappear, due to lack of oxygen, and are replaced by other species more adapted to anaerobic environments. At deeper levels, the low temperatures and the absence of oxygen and light constitute an unviable environment for life of the vernacular living beings [76].

Cyanobacteria produce toxins that are poisonous to certain living organisms, initiating the poisoning of the various links in the food chain until they reach human consumption [77]. Some of the adverse health effects are damage to the liver, intestines, and cardiovascular and central nervous systems [78–82]. In addition, cyanobacteria release various odorous substances that transform natural ecosystems into foul smelling areas [79,83,84].

The most common toxin-producing species in cyanobacteria include *Anabaena circinalis*, *Aphanizomenon flos-aquae*, *Microcystis aeruginosa*, and *Cylindrospermopsis raciborskii* [85,86]. Microcystins have been internationally assessed by the World Health Organization as cyanobacterial toxins for their health risk, which can cause acute liver injury and constitute active tumor promoters [67]. Europa, Oceania, and certain South American countries have integrated, into their national legislation, a threshold of microcystin-LR concentration in drinking water of 1 µg/L [87]. Many inland and coastal waters bodies worldwide are polluted by eutrophication [88] and life in certain seas is compromised. Since 1950, dead zones have been reported from more than 500 systems in coastal and inland water, affecting a total area of more than 300,000 square kilometers [89–91] Carpenter and Bennett [29] mention that wetland, lakes, and freshwater maximum tolerable limit of inputs nutrient receiving has already been crossed.

### 3. Current Status of the Impacts of Excessive N Fertilization

The analysis of historical trends in N fertilization seems to follow the environmental Kuznets curve (EKC) with a phase where N pollution increases and, later, decreases as economic development grows [92]. The EKC curve (bell-shaped) explains why developing countries remain in the phase of increasing pollution (e.g., India), certain countries are in the transition phase (e.g., China), while others are in a phase of reducing pollution (e.g., The USA and the EU).

#### 3.1. Developed Economies

According Zhang et al. [92], the EU appears to have reached its turning point in the late 1980s, due to changes in the EU Common Agricultural Policy (reduced crop subsidies, and EU Nitrates Directive). On the other hand, the USA supports volunteer approaches and improvement has been modest since the 1990s, based on increased productivity (maintaining fertilizer use, increasing yields) and public and private efforts to improve NUE voluntarily. In 2017, the USA emitted to the atmosphere about 8% of the N<sub>2</sub>O emitted from N fertilization worldwide [12].

The EU approach is focused on best practices and nutrient balance and promotes the adoption of nutrient management planning (NMP); however, the rate of adoption of NMP by farmers remains slow compared with other world regions such as the USA or some South American countries [93]. Overall, the N loss from farmland to the environment is expressed in terms of N balance, which decreased from 2000 to 2015 and has been stable since 2015 [94]; however, 40% of surface waterbodies, and more than 50% of groundwater, in the EU were affected in 2015 by nutrient pollution, largely due to N fertilization and manure mismanagement. The nitrogen fertilization balance is checked by EU Member States but good practices and NMP are mainly verified through document examination without any soil analysis [95]. Several socio-economic variables (farm size, farmer age, etc.) have been suggested to explain the low adoption of NMP. Daxini et al. [96] analyze the factors behind the low adoption rate and conclude that farmer psychology needs to be better understood, in relation to the use of management practices, to provide both environmental and financial benefits.

#### 3.2. Transition Economies

Developing countries, where 76% of the world population lives, use more N than developed countries. Asia is the greatest consumer of N fertilizer and energy in the world and is consequently responsible for a considerable proportion of the global creation of N<sub>2</sub>O. Approximately 58% of N is currently emitted into the environment as a pollutant [12].

China is the second-largest country in the world with a cultivated area of 164 million ha, of which 63 million ha are irrigated land. Boosted by the policy goal of achieving a 95% self-sufficiency rate, the Chinese government provides large amounts of subsidies for agricultural activity, including fertilization. Consequently, China is the world's largest fertilizer consumer with an average annual growth of 5.2% over the last three decades (2.5 times

the EU growth rate) and accounts for more than one third of the chemical fertilizers consumption worldwide (equivalent to the USA and India consumption combined). [97]. The fertilizer used rate (kg/ha) in China is 2.8 times as that of the USA and 3.0 times as that of the world average [98]. In 2017, China used 29.55 million tons of N fertilizers, and emitted a total of 1275 gigagrams of  $N_2O$  into the atmosphere [12]. Therefore, China urgently needs to formulate and implement effective measures to reduce greenhouse gas emissions from the agricultural sector [99].

Regarding eutrophication caused by leaching, there are multiple observations of high concentrations of cyanobacterial blooms in the Chinese river network (e.g., Xiang et al. [100]; Feng et al. [101]; 60% of China's underground water is unfit for human consumption [102].

India, with 26.9 million tons of N fertilizer used in 2017, emitted 722.76 gigagrams of  $N_2O$  into the atmosphere, and is the second-largest  $N_2O$  emitter in the world after China [12]. Recently, evidence of high anthropogenic-induced N and phosphorus concentrations has been found in the Ganges basin, especially from Ramganga and Kali rivers. [103].

Despite the concerning situation in China and India, there are promising initiatives to reduce pollution from agricultural fertilizers. The Chinese government's 2015 action plan to avoid fertilizer use or the Indian government's support for the "Zero Budget Natural Agriculture" project to provide financial assistance to farmers, could prove to be good examples [104].

South and Central America recorded the largest increase in N use in agriculture over the last 10 years [12]. This is largely due to the deforestation that is taking place in the Brazilian Amazon to increase the agricultural land, which currently accounts for 28.8% of Brazil's surface area [105]. Brazil is the country with the largest volume of freshwater reserves in the world. However, the high level of water pollution in Brazil means that many parts of the country lack the drinking water necessary to meet basic needs [106–108], and agricultural water runoff with high concentration of nutrients and toxic elements is the cause of this level of pollution [109–111]. Toxic cyanobacterial blooms have been observed in certain lagoons on the coast of Brazil [85,86,112,113], in the mouth of its rivers [114], in its reservoirs [115,116], and in other waterbodies [117]. On the other hand, the N used by agriculture in Brazil during 2017 was the cause of the emission of 574.86 gigagrams of  $N_2O$  into the atmosphere (7.41% of the  $N_2O$  emitted worldwide). The rate of  $NH_3$  volatilization increases due to the fast hydrolysis of urea caused by high levels of tropical soil moisture and temperature [118]. This makes Brazil the fourth-largest emitter in the world of this harmful greenhouse gas, which is seriously affecting climate change, only behind China, India, and the USA [12]. Argentina is the second country in South America in terms of  $N_2O$  emissions into the atmosphere with 152.93 g in 2017 (almost 2% of the  $N_2O$  emitted worldwide) [12,119]. Abril et al. [120] detected excessive doses of N fertilizers applied to wheat crops in the wet pampas that caused the leaching contamination of groundwater. The association between high groundwater levels and intense fertigation activity in agriculture has been verified in southeast Buenos Aires [121]. In the undulating Pampas, 40% of the assessed wells with access to the Pampean aquifer contained nitrate concentrations above the maximum recommended threshold for human consumption (45 ppm) [122].

### 3.3. Developing Economies

In Africa, due to the scarcity of surface water, potential socio-economic development relies on groundwater. The two main dangers in the use of groundwater are overexploitation and pollution of available reserves [123]. According to Usher [124], groundwater degradation is the most serious water-resource problem in Africa, and nitrate pollution is considered the major groundwater pollution issue. Africa, as a whole, has seen an increase in N fertilizer use of around 46% over the period 2007–2017 [12].

Egypt utilizes 30% of all the N used in Africa in fertilization. The bulk of Egyptian agriculture is based in the Nile Delta, which is one of the largest river deltas in the world. Recently, Hegazy et al. [125] have found a concentration of nitrates ( $NO_3^-$ ) and ammonium ( $NH_4^+$ ) in the Nile Delta, of 25 mg/L and 3 mg/L, respectively. Nigeria, due to the



continuous expansion of its arable land, is the third-largest consumer of N fertilizers in Africa, only behind of Egypt and South Africa [12]. Nigeria has increased its arable land share of the country from 51.84% in 1981 to 77.36% in 2016 [98], and this has been accompanied by another increase in the use of fertilizers, which, in 2017, was 456 million tons with an estimated emission of N<sub>2</sub>O of 106.37 gigagrams [12].

#### 4. Nutrient Use Efficiency

Nitrogen use efficiency is a term used generically in the agronomic literature. Nutrient use efficiency refers not only to the ability of crops to absorb nutrients, but also to their capacity to be used, and can therefore be defined in terms of production outputs relative to nutrient inputs or in terms of recovery of applied nutrients [126]. In this review, we find estimates made using agronomic, physiological, or economic criteria; calculations of N recovery based on measurements of grain quantity or grain plus biomass; data on N used to include N fixed in the soil from previous crops or not.

Partial Factor Productivity (PFP) analyzes the crop yield per unit of applied N. It is a useful indicator for farmers because of its easy calculation. However, for macro analysis it presents the difficulty that PFPs vary between crops, since each crop has different nutrient and water requirements. Fixen et al. [127] estimated a world average PFP of 44 kg of cereal per kg of N applied after analyzing the NUE of 25 consecutive years (1983–2007). Data for the period analyzed showed a slight increase in the NUE at the global level, driven by countries with developed economies, such as the USA, Germany, the United Kingdom, and Japan, with partial maximum factor productivity in Sub-Saharan Africa and minimum in East Asia. In general, the historical evolution shows an upward trend in the productivity of N used in Africa, the USA, and Europe, while in India, Latin America, and China it has been steadily decreasing.

Several studies show that the NUE is around 55% worldwide. Ladha et al. [128] conducted an extensive analysis based on 93 studies obtaining a final result of 55% RE, which coincides with the estimate of 57% global NUE published by Sheldrick et al. [129] or with the official estimates from the USA of 56% [130].

Recent studies estimate a global NUE of 43% [131]. Zhang et al. [92] estimated that food security and environmental protection goals will not be achieved until the global average of the NUE increases from the current ~0.4 to ~0.7 (target 2050). The NUE obtained from research plots is often lower than that from farm fields [132]. Generally, the causes of low NUE are focused on technical aspects. However, a certain inefficiency may also be explained by socio-economic factors.

An increase in NUE rates implies a rational management of nutrients by the farmer in order to increase the efficiency of his resources, being more respectful with the environment while improving his economic profitability [133]. Many crop management practices can help reduce pollution from N fertilizers. The International Plant Nutrition Institute, the International Fertilizer Industry Association, the Fertilizer Institute, and the Canadian Fertilizer Institute have developed a framework for sustainable fertilizer management called “4R Nutrient Management”. This framework is being implemented worldwide and synthesizes good practices in the field of fertilization in using the right nutrient source, applied at the right rate, at the right time, and in the right place, to achieve improved sustainability. These techniques are the basis of the Best Management Practice (BMP) that also include other practices aimed at reducing diffuse pollution such as buffer strips capable of retaining the N concentrations not absorbed by crops [134]; cover crops [135]; use of varieties that are genetically more efficient in the use of N [136]; increasing in appropriate crop rotations [137]; soil tillage systems [138]; fertilization strategies based on soil analysis [139]; methods based on N analysis in plant analysis [140] or dynamic computer simulation and decision support system [141].

## 5. Policies for the Sustainable Management of N

The efficiency of all these strategies and techniques has been widely proven to increase the NUE of crops and therefore, to reduce diffuse N pollution caused by the agricultural sector. However, these BMPs have not been widely accepted by farmers who encounter several barriers to changing their management practices [142,143]. The process of acceptance and implementation of BMPs by farmers is influenced by factors such as government policy, the psychological characteristics of the farmer, and the economic and agronomic characteristics of the farm [144,145].

Understanding the decision-making process by farmers is fundamental to improving fertilizer management [131]. One of the psychological factors that represent a major obstacle to the adoption of these new practices is the widespread skepticism among farmers about climate change and the link between N fertilizers and climate change. An effort must be made to educate and publicize the risk of environmental pollution caused by N fertilization, the fatal consequences that we can already see today, and that have been analyzed above. Similarly, empowering farmers as a key element in the fight against climate change would minimize negative reactions [142]. Public policies aimed at reducing N pollution will not be fully successful until they are aligned with the mechanisms that motivate farmers to change their management practices. Emphasis should be placed on strengthening positive attitudes that highlight the benefits of adopting these BMPs [96].

### 5.1. Policies

Environmental policies in general aim to internalize, prevent, and correct the externalities of economic activities. Regarding N use as a fertilizer, the externalities are clearly stated in the previous sections and the policies are aimed at changing farmers' management practices. Each country has its own system of environmental regulation with different levels of requirements for farmers. Policies can be categorized broadly as: (a) mandatory measures, such as quotas, technical requirements, calendars, technologies; (b) voluntary and incentive-based measures, such as European CAP Eco schemes and cross compliance or EU Rural Development Funds; (c) awareness-raising and private initiatives.

Kanter et al. [131] propose the concept of a complete food chain NUE, as a new framework for analysing policies to reduce N pollution. In their study, they show that, given the close influence between all the actors in the food chain, N losses at farm level can be reduced by applying policies aimed at non-farmers.

The key policy instruments for diffuse N pollution in the EU are provided by the Nitrates Directive that was subsequently complemented by the Water Framework Directive [146]. Optimization of farmer management practices regarding fertilizer remains an objective of the CAP, which has included environmental sustainability since the reform in 2000 and defines monitoring as a central measure for Cross compliance under the PAC subsidies scheme. The result of monitoring shows that compliance with the Sludge Directive and the Habitat Directive was good but the N Directive was found to be deficient in various MS (EEA, 2020). In the USA, there are some fertilizer management regulations (manure and nutrient management planning or phosphorous index), but almost no federal standards related to agricultural fertilizer management are mandatory, the most of them are a system of voluntary programs.  $\text{NH}_x$  emissions in the USA remain largely unregulated [147]. In China, the current regulation on N pollution is very weak. However, the high level of air and water pollution in many Chinese regions has forced the Chinese Ministry of Agriculture to initiate a reform of the subsidy system based on ecological principles [102].

### 5.2. Social Intervention

The main proposals for the reduction in excess nitrates involve a limitation of the total amount of N added per ha. World institutions such as OECD and EU endorse nutrient budgets as an environment indicator and promote the use of farm quotas. Farmers in the European Union are subject to the Nitrates Directive and the Emissions Ceiling

Directive [131] although the review of current legislation in Europe by [148], finds that N budgets are obligatory only in Switzerland, Germany, and Romania.

The development of an inventory of existing and potential policies worldwide to improve agricultural N management would involve a vast task that exceeds the objective of this study, as each country has its own regulatory system with different levels of stringency. For example, in the USA, agricultural policy is based on voluntary programs such as the Environmental Quality Incentive Program. Regarding China, the most prolific user of N fertilizers worldwide, there is growing environmental awareness that demands improvement, and this is being matched by high-level policies, such as the renewed National Environmental Protection Law which establishes highly punitive sanctions for polluting companies that fail to comply with environmental regulations [149].

### 5.3. Economic Instruments

Regarding taxes, economists have argued that the price of fertilizer would have to be doubled in order to induce fertilizer savings [150], and the analysis of the EU experience with fertilizer taxation suggests that, overall, the effectiveness of pricing remains limited [151].

In terms of economic tools, the use of tradable nutrient-reduction permits has been suggested for the Baltic Sea Action Plan [152]. However, these policies are difficult to implement, and farmers are only one of several types of actors involved in the agri-food chain.

Alternative instruments for the improvement of EU N Directive pollution control have recently been analyzed by several authors, such as [153], who study alternatives for the reduction of nutrient surplus (N) on dairy farms, and conclude that manure application standards would be more effective in limiting nutrient surplus to soils compared to the use of nutrient surplus tax.

### 5.4. Technical Innovations

The dominant approach towards improving NUE is technological and can be defined by the '4Rs' label: applying the right supply, at the right rate, at the right time, in the right place. In addition to the 4R concept promoted in the USA, effective practices include adaptive management and on-farm research where farmers are organized locally to collect data on their fields and optimize their N fertilizer management. This strategic approach is complemented with research and innovation in the fields of plant breeding, irrigation, and agronomic practice. New technological developments are also needed, such as slow-release and controlled-release fertilizers, whose aim is to release N as the crop requires, thereby reducing the loss of applied N fertilizer and increasing the NUE. Additionally, certain authors promote the use of nitrification and urease inhibitors that are now widely applied in agriculture [154].

Precision irrigation and drip irrigation also constitute a growing trend promoted by many governments [155]. The implementation of water conservation technologies also facilitates the use of fertigation (the application of fertilizer via irrigation water).

Chen et al. [102] show that subsidizing innovation is a valid tool for reducing emissions from agriculture. While production subsidies are not environmentally friendly (e.g., China), innovation subsidies decrease the total emission of pollutants while helping to increase the farmer's margin.

## 6. Conclusions

Diffuse pollution, caused by N fertilization in agriculture, has become a global problem that is difficult to solve. Through a systematic review of published documents, we have analyzed the relationship between N fertilizer use, NUE, no-point pollution, policies, and farmer management practices. The results show that the trend in N fertilizer use worldwide continues increasing. Only the most economically developed countries (EU and the USA) are starting to moderate their growth rate of N use in agriculture in relation



to their crop yields. In addition to this level of fertilization, studies show that we continue with a very low NUE (around 40% worldwide), leaving around 60% of N used as a potential environmental pollutant. Numerous studies show the direct relationship between agricultural N fertilization and diffuse pollution through volatilization, leaching and runoff. The key to reduce this pollution is to increase the NUE. There is also a lot of literature focused on describing, developing, and demonstrating the effectiveness of techniques and procedures that help to improve NUE and reduce its negative effects both locally and globally. Some of these practices have been incorporated into the legislation of some countries (mainly in the European Union). However, these standards have not been generally accepted by farmers, who are reluctant to put them into practice. Existing studies in this area link awareness, willingness, and management change. Some studies report on the defensive attitude of farmers promoted by the lack of information on the reality and magnitude of the problem caused by N fertilization.

The complexity, and current level, of diffuse pollution suggests a better understanding of the factors that influence the management practices of farmers. This will lead to more effective policies that can be designed and applied to implement BMPs in a generalized way and, ultimately, reduce diffuse N pollution through the improvement of the NUE.

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